Title: HIGH EFFICIENCY DESULFURIZATION OF SYNTHESIS

GAS: II

Authors: Douglas P. Harrison, Co-Principal Investigator

Elizabeth J. Podlaha, C0-Principal Investigator

Kwong-Bok Yi, Graduate Assistant

Louisiana State University

Gordon A. and Mary Cain Department of Chemical Engineering

Baton Rouge, LA 70803
Tel: 225-578-3066
Fax: 225-578-1476
E-mail:harrison@che.lsu.edu

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ABSTRACT

A previous study has shown that reduced cerium oxide, CeO_n with 1.5 < n < 2.0, was capable of reducing the H_2S concentration to 1 ppmv or less at temperatures in the range of $650^{\circ}C$ to $800^{\circ}C$. This is a considerable improvement over the performance of current generation zinc-based sorbents at these conditions, and shows promise for achieving Vision 21goals of reducing the H_2S in coal gasification process streams to sub-ppmv levels. In addition, the sulfided product, Ce_2O_2S , can be easily regenerated with the direct formation of elemental sulfur, thereby avoiding the need for further treatment of the regeneration product gas to prevent atmospheric emissions of SO_2 .

Because of the stability of CeO₂, an oxygen-free, high H₂ gas was required to achieve reduction. Gases produced from typical gasifiers such as Texaco and KRW do not have the reducing power to form appreciable amounts of CeO_n. Recent research involving three-way automotive and oxidation catalysts has shown that the catalytic performance of CeO₂ is enhanced by the addition of ZrO₂. It is said that, among others, ZrO₂ increases the oxygen reducibility and improves the thermal stability, factors that should also improve the performance of ceria-based desulfurization sorbents in less reducing atmospheres.

CeO₂-ZrO₂, available commercially and synthesized in-house electrochemically are being studied. In-house synthesis provides the flexibility of altering the sorbent structure and composition for optimal sulfur removal. In-house synthesis and characterization of CeO₂-ZrO₂ materials were reported during the 2001 UCR meeting.

Experimental effort during the past year has used a laboratory-scale fixed-bed reactor to determine H_2S removal efficiency as a function of temperature and sorbent and feed gas composition. The reactor and feed and product lines are constructed of quartz, teflon, and silcosteel to eliminate interaction between H_2S and stainless steel surfaces. Product gas concentration is measured by gas chromatography using a pulsed flame photometric detector (PFPD) for low H_2S levels (0.1 to 10 ppmv) and a thermal conductivity detector (TCD) for higher concentrations.

Journal Articles

Electrochemical Synthesis of Nanocrystalline Ceria-Zirconia, A. Mukherjee, D. P. Harrison, and E. J. Podlaha, Electrochemical and Solid State Letters, 4, D5, 2002

Presentations

High Temperature Desulfurization of Synthesis Gas: I., A. Mukherjee, K. B. Yi, E. J. Podlaha, and D. P. Harrison, presented at the DOE University Coal Research Conference, Pittsburgh, June 2001.

Electrodeposition of Ce-Zr Powder, A. Mukherjee, D. P. Harrison and E. J. Podlaha, presented at the Joint International Meeting of the Electrochemical Society, San Francisco, September 2001.

Electrochemical Synthesis and Characterization of $Ce_{1-x}Zr_xO2$ Nanocrystalline Powders, E. J. Podlaha, J. Chan, J. Jiang, A. Mukherjee, and D. P. Harrison, presented at the Annual Meeting AIChE, Reno, November 2001

Ceria-Zirconia High Temperature Desulfurization Sorbents, K. B. Yi, E. J. Podlaha, and D. P. Harrison, presented at the Annual Meeting AIChE, Reno, November 2001.

Students Receiving Support from the Grant

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